Prediction of Technetium-99m Yield from Molybdenum-99 Generators

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The prediction of the formation of a daughter product as the result of radioactive decay in a generator is complicated by the quantity of daughter product remaining in the generator after the last elution. Good laboratory practice indicates that there should be a method of predicting yield for comparison with on-site assay results.

The methods of computation are not difficult, but they can be much simplified for routine application. To describe the method it is necessary to begin with some basic concepts.

The decay scheme of 99 Mo may be diagrammed as in Fig. 1. The percentages of the various branches are according to Crowther and Eldridge (1).

Because of the specific interest in the metastable state of Technetium-99, the pathway of interest is indicated in the above figure by the numerals I, II, and III.

The rates of disintegration for these three decay modes are given by (2):

$$dN_I = -\lambda_I N_I$$

$$\frac{dN_{II}}{dt} = \lambda_I N_I - \lambda_{II} N_{II}$$

$$\frac{dN_{III}}{dt} = \lambda_{II}N_{II}$$

Where N is the number of atoms present, λ is the decay constant, and t is the elapsed time.

Using A to denote decay rate (Curiage), of the components, these equations may be expressed:

4.
$$A_{I} = A_{I}^{\circ} \exp(-\lambda_{I} t)$$

5.
$$A_{II} = \frac{A_I^{\circ}}{\lambda_{II} - \lambda_I} \left[exp(-\lambda_I t) - exp(-\lambda_{II} t) \right] + A_{II}^{\circ} exp(-\lambda_{II} t)$$

 A° denotes the activity of each nuclide present when t=0. Because of the very long half-life, the amount of 99-Ru formed will be inconsequential and the equation for this transformation has been omitted. The last factor in Equation 5 expresses the residual 99m Tc in the generator after the last elution.

Only 87.6% of the ⁹⁹Mo decays to ^{99m}Tc and the equation must be corrected for this factor.

The conditions at transient equilibrium (t > 44 hours) is:

Figure two is a semilogarithmic plot of the decay of ⁹⁹Mo and the formation of ^{99m}Tc after total and incomplete elutions. It can be seen that the amount of ^{99m}Tc available will be affected by the elution and should be considered in the calculation of the available quantity of technetium-99m.

In the practice of generator operation there are two conditions under which prediction may be desired: more than 44 hours since the last elution (transient equilibrium) and less than 44 hours since the last elution.

CASE I. More than 44 hours since the last elution.

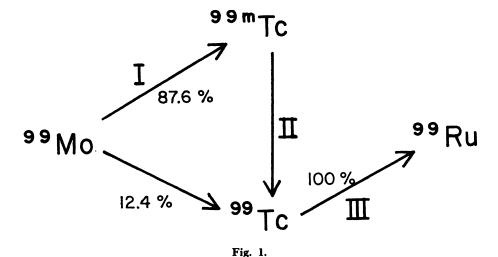
In this case (transient equilibrium) the calculations are simple. Calculate the activity of the ⁹⁹Mo at the time of elution by decay computation from the manufacturer's assay.

The total Technetium available will be 96.4% of the Molybdenum activity. From the laboratory assay of the eluted 99mTc, calculate the elution efficiency. CASE II. Less than 44 hours since the last elution.

For this case a solution to Equation 6 is required. The electronic digital computer has been used to prepare a table of solutions.

Table I is a table of the decay of 99Mo and formation of 99mTc following elution for various prior elution efficiencies from 40% to 90%. The table is used as follows:

From the elapsed time since the last elution and the ⁹⁹Mo activity at that time, compute the current ⁹⁹Mo activity. The entry in the table opposite the elapsed time in the column representing the last elution efficiency is the ratio



of 99m Tc to 99 Mo activity. This factor multiplied by the 99 Mo activity yields the value of the total available technetium-99m.

SAMPLE PROBLEM.

At t = 0, $^{99}Mo = 100$ mc. ^{99m}Tc yield = 57.6 mc, or 60% elution efficiency. At t = 24 hours, $^{99}Mo = (100) (0.780) = 78$ mc. $^{99m}Tc = (78) (0.91) = 71$ mc.

TABLE I. 99m-Tc/99-Mo

Time since last elution	99- Mo (N/No)	0.401	0.501	0.601	0.701	0.80 ¹	0.901
0	1.000	0.53	0.44	0.35	0.26	0.18	0.09
4	0.960	0. 6 8	0.62	0.56	0.50	0.44	0.38
8	0.921	0.77	0.73	0.70	0.66	0.62	0.58
12	0.883	0.84	0.81	0.70	0.76	0.73	0.71
16	0.847	0.88	0.86	0.85	0.83	0.81	0.80
20	0.813	0.91	0.90	0.89	0.87	0.86	0.85
24	0.780	0.93	0.92	0.91	0.91	0.90	0.89
28	0.748	0.94	0.94	0.93	0.93	0.92	0.92
32	0.717	0.95	0.95	0.94	0.94	0.94	0.93
36	0.688	0.95	0.95	0.95	0.95	0.95	0.94
40	0.660	0.96	0.96	0.95	0.95	0.95	0.95
44	0.633	0.96	0.96	0.96	0.96	0.96	0.96
48	0.607	0.96	0.96	0.96	0.96	0.96	0.96

¹Elution Efficiencies

TABLE II. 99m-Tc RELATIVE YIELD

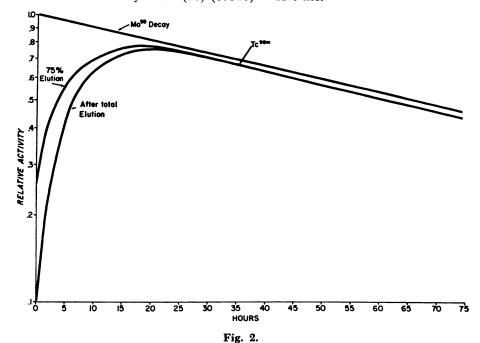
Time since last elution	99-Mo (N/No)	0.401	0.501	0.601	0.701	0.801	0.901
0	1.000	0.212	0.220	0.210	0.182	0.144	0.081
4	0.960	0.272	0.310	0.336	0.360	0.352	0.342
8	0.921	0.308	0.365	0.420	0.462	0.496	0.522
12	0.883	0.336	0.405	0.474	0.532	0.584	0.639
16	0.847	0.352	0.430	0.510	0.531	0.648	0.720
20	0.813	0.364	0.450	0.534	0.609	0.688	0.765
24	0.780	0.372	0.460	0.546	0.637	0.720	0.810
28	0.748	0.376	0.470	0.558	0.651	0.736	0.828
32	0.717	0.380	0.475	0.564	0.658	0.752	0.837
36	0.688	0.380	0.475	0.570	0.665	0.760	0.846
40	0.660	0.384	0.480	0.570	0.665	0.760	0.855
44	0.633	0.384	0.480	0.576	0.672	0.768	0.864
48	0.607	0.384	0.480	0.576	0.672	0.768	0.864

¹Elution Efficiencies

If the elution efficiency is still 60%, the yield will be 60% of 71 mc, or 42.6 mc. Table II is useful if the generator operation is sufficiently controlled to yield a relatively constant elution efficiency. This table is used in the same way as Table I, except that the last two computations are combined.

SAMPLE PROBLEM.

At t = 0, $^{99}Mo = 100$ mc. ^{99m}Tc yield = 57.6 mc, or 60% elution efficiency. At t = 24 hours, $^{99}Mo = (100) (0.780) = 78$ mc. ^{99m}Tc yield = (78) (0.546) = 42.6 mc.



SUMMARY

In the laboratory operation of the ⁹⁹Mo/^{99m}Tc generators, it is customary to assay the eluant for specific activity and total activity of ^{99m}Tc pertechnetate. A simple method is described which enables the user to compute the available ^{99m}Tc at any time following elution.

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