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RADIOLANTHANIDES AS PROMISING TUMOR SCANNING AGENTS

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Based on the previous findings that some of the elements in the Group III and in the Period VI of the periodic table have a tumor affinity, the presence and extent of tumor affinity of radiolanthanides were examined using Yoshida sarcoma-bearing rats. Results indicated that all eight elements tested show some affinity for a tumor. Thulium-170 and ¹⁶⁹Yb are nuclides that can favorably compare with that of ⁶⁷Ga. Discovery of other good nuclides was discussed.

As a promising nuclide for clinical tumor scanning, ¹⁶⁷Tm, ¹⁶⁹Yb, and ¹⁷⁵Yb (citrate form is preferable) were suggested.

Since our success was reported in 1965 (1) on positive delineation of human tumors with ¹³¹Ihuman serum albumin, the search for a good tumorselective compound has been continued in our laboratory. Fairly good results of clinical tumor scanning were also obtained with ^{99m}Tc-albumin in 1966 (2), ²⁰³Hg-hematoporphyrin in 1966 (2,3), and ¹³¹Ifibrinogen in 1968 (4). Recent successes with ⁶⁷Gacitrate by Edwards and Hayes in 1969 (5) and with ¹¹¹In-chloride by Hunter and Riccobono in 1970 (6) have converted our interest to simple compounds.

A series of basic experiments has been carried out on Yoshida sarcoma-bearing rats using many elements in the form of citrate, chloride, or nitrate. Elements tested so far in our laboratory number 28, including phosphorus, calcium, scandium, vanadium, iron, zinc, gallium, germanium, arsenic, selenium, strontium, yttrium, zirconium, niobium, technetium, silver, cadmium, indium, tin, antimony, iodine, cesium, barium, hafnium, tantalum, gold, mercury, and bismuth. The results have been summarized as follows from the viewpoint of tumor affinity (7): gallium and indium in Group III showed a very strong affinity to tumor, and scandium also showed a slight affinity to Yoshida sarcoma; mercury and bismuth in Period VI had a very strong affinity to tumor, and gold also had a very strong affinity to tumor when used as a H¹⁹⁸AuCl₄ solution. Considering these facts carefully in relation to the periodic table, it seems logical to select thallium and the lanthanides as key elements. The ²⁰²Tl-citrate did not show any affinity to Yoshida sarcoma but some of the lanthanides, such as ¹⁷⁰Tm, ¹⁶⁹Yb, and ¹⁷⁷Lu, showed excellent tumor affinity, true to our expectations.

MATERIALS AND METHODS

Among 15 lanthanides, 8 radioactive nuclides were available: ¹⁴⁰La, ¹⁴¹Ce, ¹⁵³Sm, ¹⁵³Gd, ¹⁶⁰Tb, ¹⁷⁰Tm, ¹⁶⁹Yb, and ¹⁷⁷Lu. All nuclides but ¹⁴⁰La (chloride) were used as citrate. Donryu rats each weighing 150-200 gm underwent subcutaneous implantation of Yoshida sarcoma (2 \times 10⁸ cells/0.5 ml) in the right thigh. Six to 7 days later an appropriate amount of radioactive nuclide was injected intravenously through the tail vein, at which time the tumor had grown to 2 cm in diam. The amounts administered to a rat were ¹⁴⁰La 3 μ Ci/lanthanum 40 μ g; ¹⁴¹Ce 2 μ Ci/cesium 7.2 μ g; ¹⁵³Sm 3 μ Ci/ samarium 36 μ g; ¹⁵³Gd 2 μ Ci/gadolinium 0.006 μ g; ¹⁶⁰Tb 2 μ Ci/terbium 5.4 μ g; ¹⁷⁰Tm 33 μ Ci/thulium $\frac{3}{2}$ µg; ¹⁶⁹Yb 2 µCi/ytterbium 11 µg; and ¹⁷⁷Lu 30 μ Ci/lutetium 3 μ g, respectively. Five animals were killed 3, 24, and 48 hr postinjection, respectively. Tumor and organ specimens obtained at autopsy were assayed by a well scintillation counter. The

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retention values (percent of administered dose per gm/tissue weight) in various organs were calculated.

In the second experiment, to compare the distributions of ⁶⁷Ga and ¹⁶⁹Yb ions in vivo more precisely without any individual variation and tumor growth difference, we administered 67Ga and 169Yb simultaneously as a mixture of the same chemical form of citrate to the Yoshida sarcoma-bearing Donryu rats. The rats were killed at 3, 24, 48, and 72 hr following intravenous injection. Separate assessment of two nuclides was made using the difference of decay speeds of both nuclides with a half-life of ¹⁶⁹Yb 32 days, ⁶⁷Ga 78 hr. It was quite easy to compute the amount of each of the two nuclides after determination at two times: immediately and 12 days after autopsy.

RESULTS

Table 1 shows the retention values (%/gm) of each radionuclide in the tumor at 3, 24, and 48 hr after injection. Data for ⁶⁷Ga-citrate are shown in the last line for comparison. All radionuclides show more or less affinity to the malignant tumor, and the retention value in the tumor tissue of ¹⁷⁰Tmcitrate is the highest-1.34%/gm-and is followed by ¹⁶⁹Yb-citrate and ¹⁷⁷Lu-citrate. Generally, the lanthanides with higher atomic number have stronger affinity to a tumor than the ones with lower atomic

number such as ¹⁴⁰La, ¹⁴¹Ce, and ¹⁵³Sm. Results in the carefully designed second experiments (Fig. 1) showed that the retention value of ¹⁶⁹Yb-citrate in the tumor tissue itself was similar to that of ⁶⁷Gacitrate. The great difference was noticed in the distribution in normal tissue between ¹⁶⁹Yb and ⁶⁷Ga. The ¹⁶⁹Yb-citrate is cleared from the peripheral blood more rapidly than ⁶⁷Ga-citrate. Retention values of ¹⁶⁹Yb-citrate in the liver and spleen were less than those of ⁶⁷Ga-citrate whereas accumulation of ¹⁶⁹Yb in the bone was almost two times that of ⁶⁷Ga. These differences may cause the lower body background in soft tissues and the denser bony image on the scintigram in the case of ¹⁶⁹Yb-citrate.

DISCUSSION

In 8 of 15 lanthanide elements, the presence and extent of tumor affinity were examined. Thulium, ytterbium, and lutetium were found to have a relatively strong affinity for tumor that was no less than that of ⁶⁷Ga-citrate. There still remains the possibility that another tumor-selective element will be discovered among the remaining seven elements that were not available for us at present. The possibility might be great particularly in erbium, holmium, and dysprosium, which have a relatively large atomic number.

As a possible scanning agent, thulium has many gamma-emitting nuclides: ¹⁶⁵Tm, ¹⁶⁶Tm, ¹⁶⁷Tm,

	Time (hr)	Tumor	Blood	Muscle	Liver	Kidney	Spleen	Bone
	3	0.52	2.20	0.035	4.90	0.42	0.59	
¹⁴⁰ La-chlorid e	24	0.36	0.035	0.009	7.21	0.64	0.67	
	48	0.31	0.010	0.010	6.29	0.97	1.00	
¹⁴¹ Ce-citrate	24	0.38	0.016	0.025	10.23	1.26	0.54	0.55
	3	0.66	1.90	0.18	7.11	1.75	0.79	0.28
¹⁶⁸ Sm-citrate	24	0.44	0.022	0.043	8.53	2.38	0.40	0.78
	48	0.45	0.008	0.052	8.05	2.54	0.23	1.04
¹⁵³ Gd-citrate	24	0.37	0.018	0.020	2.12	1.28	0.14	2.97
¹⁶⁰ Tb-citrate	24	0.41	0.021	0.056	1.42	2.16	0.21	2.94
	3	1.26	0.20	0.081	0.92	2.46	0.28	2.41
¹⁷⁰ Tm-citrate	24	1.34	0.014	0.025	0.53	0.71	0.18	3.04
	48	1.30	0.012	0.017	0.34	0.51	0.10	3.10
	3	0.73	0.50	0.097	0.55	1.34	0.41	1.8
¹⁰⁹ Yb-citrate	24	0.72	0.014	0.019	0.38	0.84	0.27	1.92
	48	0.73	0.013	0.019	0.44	0.74	0.29	1.75
	3	0.55	0.41	0.095	0.60	1.50	0.35	
¹⁷⁷ Lu-citrate	24	0.59	0.022	0.025	0.47	0.64	0.32	
	48	0.58	0.017	0.026	0.44	0.61	0.30	
	3	0.73	1.88	0.17	0.99	0.65	0.76	
^{e7} Ga-citrate	24	1.14	0.19	0.050	1.41	0.75	1.29	
	48	0.69	0.041	0.091	1.71	0.81	1.91	

TABLE 1. DEPOSITION OF EACH LANTHANIDE AND 67Ga IN YOSHIDA SARCOMA-BEARING

mean value of data of five animals.

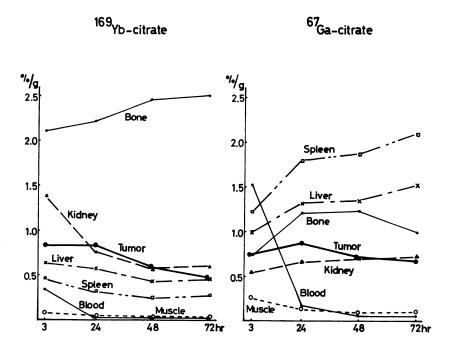


FIG. 1. Ytterbium-169-citrate and "Gacitrate were injected simultaneously into Yoshida sarcoma-bearing rats. Ordinate is retention value of radionuclide expressed as percent of administered dose per gramtissue weight in various tissues and tumor. Abscissa is time intervals after administration of ¹⁰⁰Yb and "Ga. Note great difference in distribution in normal tissue between two scanning agents.

¹⁶⁸Tm, ¹⁷⁰Tm, and ¹⁷²Tm. From the point of view of proper physical half-life, adequate energy of the principal gamma ray and no beta emission if possible, ¹⁶⁷Tm must be the nuclide of choice. It has been reported recently as a bone-scanning agent (8) but is not available yet in Japan. On the other hand, ytterbium has ¹⁶⁶Yb, ¹⁶⁷Yb, ¹⁶⁹Yb, ¹⁷⁵Yb, and ¹⁷⁷Yb as gamma-emitting nuclides and, of these, ¹⁶⁹Yb is most easily available and very popular in Japan as ¹⁶⁹Yb-DTPA for cisternography. For this reason, ¹⁶⁹Yb-citrate was used on trial in the patients with cancer. The clinical results of tumor scanning will be reported later. Ytterbium-175 is also a promising nuclide. Lutetium has ten gamma-emitting nuclides, but none of these are suitable for scanning.

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