NM/ CONCISE COMMUNICATION

SPECIFIC GAMMA-RAY AND EQUILIBRIUM ABSORBED-DOSE CONSTANTS FOR BARIUM-135m

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An examination of currently used radionuclides in bone scanning (1) would indicate the need for a better agent. Recent production and evaluations of ^{135m}Ba in animals (2-4) and humans (5) strongly suggest it to be a potential radionuclide for bone scanning among the alkaline earth metals. Its halflife of 28.7 hr and photon energy of 268 keV are most desirable for skeletal imaging. We therefore intend to present aspects of calibration and dosimetry of ^{135m}Ba before its widespread use. We also feel that there is a need for publication of specific gammaray constants (Γ) of important radionuclides in MIRD pamphlets.

Barium-135m decays to stable ¹³⁵Ba by 100% isomeric transition from an energy level of 0.268 MeV with a half-life of 28.7 hr. The 268-keV photons of ^{135m}Ba are 85% internally converted ($\alpha_{\rm K} = 3.8$, $\alpha_{\rm L} = 1.42$, and $\alpha_{\rm M} = 0.47$).

SPECIFIC GAMMA-RAY CONSTANT AND CALIBRATION

The value of Γ , the specific gamma-ray constant, must be known to calibrate the activity of ^{135m}Ba using either a standard (Type 1383A) beta-gamma ionization chamber of General Radiological Ltd., a Mediac isotope calibrator, or any allied instrument. It can be calculated in units of R-cm²/mCihr using one of the following two equations which were derived (6,7) taking 33.73 eV as the energy required to produce an ion pair.

and

$$\Gamma = 195.1 \Sigma n_i E_i \left(\mu_{a/\rho} \right)_i. \tag{2}$$

Both the equations are essentially the same. They

 $\Gamma = 1.51 \times 10^5 \Sigma n_i E_i (\mu_a)_i$

can be used depending on the accessibility of the values of $\mu_{\rm a}$ (linear energy-absorption coefficient in air per cm) and $\mu_{\rm a/\rho}$ (mass energy-absorption coefficient in air in cm²/gm). These values can be obtained from Evans (7) and Radiological Health Handbook (8). The mean number of emissions per disintegration (n_i) and the corresponding energy in MeV (E_i) can be obtained from Radiological Health Handbook (8), and Lederer et al (9). Table 1 summarizes these parameters for the calculation of Γ for ^{135m}Ba. The calculated values for Γ were 0.594 and 0.593 (R-cm²/mCi-hr) from Eqs. 1 and 2, respectively.

The value of Γ is useful for the calibration of instruments for the assay of gamma-ray emitting radionuclides. Mercury-203 can be used as a reference source for the assay of ^{135m}Ba because they have similar photon energies. The value of Γ for ²⁰³Hg was calculated to be 1.318 (R-cm²/mCi-hr) using the present method. The ratio of this Γ to that of ^{135m}Ba is 2.22. We have used this factor for the assay of ^{135m}Ba radioactivity using a commercially available radionuclide dose calibrator and an ionization chamber.

EQUILIBRIUM ABSORBED-DOSE CONSTANT AND DOSIMETRY

The values of Δ_i , the equilibrium absorbed-dose constant, for ^{135m}Ba were arrived at on the basis of information obtained from Lederer et al (9), *Radio*-

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| TABLE 1. PARAMETERS FOR CALCULATING Γ FOR $^{135\mathrm{m}}$ Ba | | | | | |
|--|----------------------|-----------------------|--|---|--|
| Radiation (i) | Photon yield (nı) | Energy in MeV (E1) | $(\mu_{*})_{1}(cm^{-1}) \times 10^{5}$ | (μ _{*/ρ}):(cm ² /gm) | |
| Gamma-1 | 0.1494 | 0.2680 | 3.67 | 0.0280 | |
| Kα-1 x-ray | 0.2691 | 0.0322 | 15.77 | 0.1220 | |
| Κ α-2 x-ray | 0.1397 | 0.0318 | 16.29 | 0.1260 | |
| K β-1 x-ray | 0.0759 | 0.0364 | 11.14 | 0.0862 | |
| K β-2 x-ray | 0.0161 | 0.0374 | 10.14 | 0.0805 | |

(1)

| Radiation (i) | Mean number per disintegration (n ₁) | Mean energy (MeV) (Ē1) | ∆ı (gm-rad/µCi-hr) |
|-------------------------------|--|---------------------------|-----------------------|
| Gamma-1 | 0.1494 | 0.2680 | 0.0853 |
| K int. con. electron, gamma-1 | 0.5672 | 0.2310 | 0.2790 |
| L int. con. electron, gamma-1 | 0.2127 | 0.2624 | 0.1188 |
| M int. con. electron, gamma-1 | 0.0709 | 0.2669 | 0.0403 |
| Kα-1 x-rays | 0.2691 | 0.0322 | 0.0184 |
| Kα-2 x-rays | 0.1397 | 0.0318 | 0.0094 |
| K β-1 x-rays | 0.0759 | 0.0364 | 0.0058 |
| K β-2 x-rays | 0.0161 | 0.0374 | 0.0012 |
| L x-rays | 0.1165 | 0.0045 | 0.0011 |
| KLL Auger electrons | 0.0437 | 0.0263 | 0.0024 |
| KLX Auger electrons | 0.0194 | 0.0308 | 0.0012 |
| KXY Auger electrons | 0.0033 | 0.0353 | 0.0002 |
| LMM Auger electrons | 0.6117 | 0.0034 | 0.0044 |
| MXY Auger electrons | 1.5127 | 0.0011 | 0.0035 |

logical Health Handbook (8), and the Oak Ridge National Laboratory. The nuclear parameters are presented in Table 2 in the same format as published in MIRD pamphlets for other radionuclides. A value of 0.4507 was obtained from the table for $\Sigma \Delta_i$ for the nonpenetrating radiation. An approximate value of 0.446 can be obtained without the table by multiplying 2.13 by $n_e \overline{E}_e$ (10), the average energy per disintergration deposited in the vicinity of the nuclide, formerly known as \overline{E}_6 .

Barium-135m is a bone-seeking element, and the skeleton is the critical organ. The absorbed radiation doses due to intravenously administered ionic ^{135m}Ba were calculated for the total body and the skeleton. The values of absorbed fraction (ϕ_i) were obtained from MIRD Pamphlet 5 (11). The values of $\Sigma(\Delta_i \phi_i)$ were found to be 0.5085 and 0.4842 for the total body and the skeleton, respectively. The effective half-life was taken equal to the physical half-life of ^{135m}Ba. It was assumed that 100% of the dose was distributed uniformly in the total body of a 70-kg man with 10-kg skeleton. Skeletal uptake was taken as 60% (5) and assumed to be uniformly distributed. The radiation doses were estimated to be 0.3 and 1.2 rads/mCi for the total body and the skeleton, respectively.

SUMMARY

Barium-135m, a potential radionuclide for bone scanning, has a half-life of 28.7 hr and emits 268keV photons. The specific gamma-ray constant (Γ) was calculated. Methods of source calibration have been presented. The equilibrium absorbed-dose constant has been calculated and the radiation doses are 0.3 and 1.2 rads/mCi for the total body and the skeleton, respectively.

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